

Atlantic Academy Portland



Year 11 - Unit 1 Knowledge Organiser



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English - Year II - Unit I - Community

Driving question: What is the importance of community?

			WEEK I: An In	spector Calls - Context	RAG
1	Context	Background inform	ation		
2	Social Responsibility	The idea that peop	le should act in a v	way that helps those less privileged .	
3	1912	The year the play i	is set . Society is d	ivided by class.	
4	Class system	Working class: ha Middle class: Prof Upper class: inhe	ardest jobs/least m fessionals and busin rited money, land,	oney ness owners: had money and control titles and power	
5	Women in 1912	Lives were contro	lled by their hust	ands; could not vote	
6	Welfare in 1912	Charities were vit	tal in supporting	the poor; no government help	
7	1914-1918	World War One	raised question	s about the leadership of the upper classes	
8	1939-1945	Second World W classes fought to	Var. Priestley wr gether	ote the play during this time. Millions of people from all	
9	Capitalism	Right-wing ideas th private owners fe	at favour a system or profit	in which a country's trade and industry are controlled by	
10	Socialism	Left-wing ideas whi	ich call for a more	equal sharing of wealth and power amongst all of society	
			WEEK 2: An Ir	spector Calls -BIG IDEAS	
11	Purpose of the play	Priestley argues that	at people should h	elp those less privileged.	
12	Family	Expectations of n parents in charge	niddle class familie e, children obed	s in 1912 – know their role, be content with their position, ient and unquestioning	
13	Young and Old	Older generation = old-fashioned/traditional Younger generation = challenge authority, are willing to learn, accept responsibility and change			
14	Men and Women	Start out as stered weaker	otypes but by the	end of the play the women get stronger and the men	
15	Judgement	lt doesn't matter w	ho the Inspector i	s – what is important is the lesson and who learns it	
16	Ignorance	Older generation – they view these a this, but pretend it	n- try to ignore an Is working class doesn't happen	ything troubling such as prostitution/womanizing and drinking problems , yet ironically the middle and upper classes do	
		W	EEK 3: An Insp	ector Calls -Dramatic devices	
17	Dramatic Irony	The audience kno Birling family's shor	ows more than the t-sightedness.	e characters. Priestley uses dramatic irony emphasise the	
18	Euphemism	A way of avoiding	saying something	unpleasant	
19	Religious Imagery	Used by the Inspec	tor to suggest we	have a 'sacred' duty of care towards others and links to	
20	Graphic Imagery	Used by the Inspec	tor to shock the	Birlings and the audience	
21	Colloquial language	Informal language contrast to their ol	ge used to reflect 1 Id-fashioned paren	their social status : the younger generation use slang in ts.	
22	Symbolism	Characters symbolic Edwardian society)	olise particular so o	cial groups (Mrs Birling symbolises the hypocrisy of upper	
23	Omniscience	The Inspector se	ems to be omnis e	cient – he knows everything	
24	Anadiplosis	Beginning a sentend (Example:'Because afterwards, and wh	ce/clause by repea what happened to hat happened to he	Ating the last word /s of the previous sentence/clause. her then may have determined what happened to her r afterwards may of driven her to suicide.)	
		WE	EK 4: An Inspec	tor Calls -Act I Key Quotations	
25	Quotation		Techniques	Context	

26	.6 <i>'Hard-headed businessman</i> Hard-headed practical man of business' Repeti		Repetition Alliteration	Birling full of self-importance. His focus is on his social status – he believes he knows everything	
27	'Half shy, half assertive'		Repetition Adjective	Stage direction shows Eric is not 'complete' and has potential for change	
28	'Mummy' 'Daddy'		Nouns	Childish language used by Sheila at the start of the play	
29	'Lower costs and higher prices'		Comparatives	Birling is only pleased about the possible financial gain from Sheila's engagement – capitalist views	
30	'The Germans don't want war'		Dramatic Irony	Birling on the troubles in Europe	
31	ʻunsinkable, absolu	utely unsinkable'	Repetition Dramatic Irony	Birling on the Titanic (which sank on its maiden voyage) Priestley makes Birling look short-sighted and silly	
32	ʻpinkbrighterh	arder'	Adjectives	Stage directions – lighting to foreshadow	
		_	WEEK 5: Revis	ion - Jekyll and Hyde	
33	Chapter I	Story of the Door	Hyde tramples th	e girl; Hyde has a key and cheque.	
34	Chapter 2	Search for Mr Hyde	Utterson discove	rs Jekyll's will; Dr Lanyon returns; Utterson meets Hyde.	
35	Chapter 3	Dr Jekyll was Quite at Ease	Dinner party at Jekyll's house; Utterson is worried about the will.		
36	Chapter 4	The Carew Murder Case	Sir Danvers Carew is 'clubbed' to death by Hyde; letter found on body; murder weapon is Jekyll's walking cane.		
37	Chapter 5	The Incident of the Letter	Jekyll looks unwell; Jekyll hands over a forged Hyde letter.		
			M	VEEK 6	
38	Chapter 6	The Remarkable Incident of Dr Lanyon	Dr Lanyon dies a	nd leaves a letter for Utterson.	
39	Chapter 7	The Incident at the Window	Utterson and Enf	ield see Jekyll at his window in 'abject terror and despair'.	
40	Chapter 8	The Last Night	Hyde commits su	icide; Utterson finds Jekyll's will and confession.	
41	Chapter 9	Dr Lanyon's Narrative	Lanyon's letter de	escribes how he became ill after seeing Hyde transform.	
42	Chapter 10	Henry Jekyll's Full Statement of the Case	Jekyll tells his sto	ry of how and why he created Mr Hyde.	
		WEEK	7: An Inspector	Calls -Act 2 Key Quotations	
	Quotation		Techniques	Context	
43	'Women of the to	wn'	Euphemism	Gerald – playing down womanising/prostitution and repeated by Mrs B to highlight her ignorance	
44	'A girl of that sort	,	Euphemism	Mrs B – trying to blame Eva for her own position	
45	'Mother – stop – s	top!'	Noun Repetition Exclamative	Sheila changes her use of nouns – there is a distance between her and her parents now. She is also hysterical as she has worked out Eric's role, but her mother hasn't	
		WEI	EK 8: Jekyll and H	lyde- KEY QUOTATIONS	

46	'like some	Simile						
	disconsolate	Adjective - 'discor	Adjective - 'disconsolate'					
	prisoner'	Description of Jek	Description of Jekyll. Links to duality, Victorian values.					
47	'blistered and	Verbs	/erbs					
	distained'	Symbolism						
		Description of the	e door - links to set	tting and the Urban city. Symbolic of the characters.				
48	'trampled calmly	Verb 'screaming'						
	over the child's	Adverb 'calmly'						
	body and left her	Verb - 'left'	Verb - 'left'					
	screaming on the	Imagery	Imagery					
	ground'	Juxtaposition - foreshadows						
		Hyde tramples a girl. Links to crime and violence, Physiognomy, Darwinism/Evolution.						
		Juxtaposition thro	ughout to represer	nt duality.				
		WEEK 9:	An Inspector Ca	lls - ACT 3 KEY QUOTATIONS				
49	'One Eva Smith ha	is gone – but	Symbolism	The Inspector uses common names to represent the whole				
	there are millions	and millions of	Repetition	of the working class who need help and support				
	Eva Smiths and Jol	hn S miths still						
	left with us							
50	50 'We don't live alone. we are		Pronouns	The Inspector is Priestley's mouthpiece for social				
	members of one body'		Repetition	responsibility				
51	'they will be taugh	t in fire and	Metaphor	Represents the break-down of society. Could be				
	blood and anguish	,	Religious	foreshadowing the World Wars				
	5		imagery	, , , , , , , , , , , , , , , , , , ,				

Week I: RAG					
١.	Rate of reaction	The rate at which a reaction takes place, i.e for the reactants to be used up and the products to be formed.			
2.	Catalyst	Substance which speeds up a chemical reaction , without being used up. Provides an alternative pathway for a reaction which has a lower activation energy .			
3.	Surface area	The total area of a substance across its whole surface. A powder has a larger surface area			
4	Concontration	than lumps of a substance.			
4. r		A manufaction in a bight the and desta construct to an f ormatic the manufacture to a substance in a certain volume.			
5.	Reversible reaction	A reaction in which the products can react to re-form the reactants.			
6.	Irreversible reaction	A reaction in which the products <u>do not</u> further react to make the reactants.			
7.	Hydrated	when a crystalline structure contains water molecules between its crystals.			
8.	Anhydrous	Describes a substance which does not contain water.			
9.	Closed System	A system where no matter (particles, molecules etc) can enter or leave.			
10.	Equilibrium	The point in a reversible reaction at which the forward and backward rates of			
	-	reaction are the same. The amounts of substances present remain constant.			
		Week 2:			
11.	Le Chatelier's	When a change in conditions is introduced to a system at equilibrium, the position of			
	Principle	equilibrium shifts so as to cancel out that change.			
12.	Mixture	When some elements (or compounds) mix together and do not chemically react.			
13.	Hydrocarbon	A molecule made up of carbon and hydrogen atoms only.			
14.	Fractions	Hydrocarbons separated from crude oil, which have similar boiling points.			
15.	Fractional Distillation	A way to separate liquids from a mixture by boiling off the substances at different temperatures, then condensing them back into a liquid.			
16.	Alkanes	A hydrocarbon which contains only single carbon-carbon bonds . Molecules from this			
		group have a name ending in -ane.			
17.	Alk <u>enes</u>	A hydrocarbon which contains at least one double carbon-carbon bond in its			
		structure. Molecules from this group have a name ending in - ene.			
18.	Saturated	A hydrocarbon which contains only single carbon-carbon bonds.			
19.	Unsaturated	A hydrocarbon which contains at least one double carbon-carbon bond.			
20.	Flammable	Easily ignited and able to burn rapidly.			
		Week 3:			
21.	Viscosity	The resistance of a liquid to flowing or pouring; a liquid's 'thickness'.			
22.	Volatility	The ability of a liquid to turn to a vapour form.			
23.	Oxidised	A reaction where oxygen is added to a substance/ or where electrons are lost from			
		a substance.			
24.	Complete	Where fuel burns fully with oxygen, to form carbon dioxide and water. General equation:			
25	Incomplete	Fuel + oxygen \rightarrow carbon dioxide + water Where fuel does not fully burn with oxygen General equation:			
23.	combustion	Fuel + oxygen \rightarrow carbon dioxide + water + carbon + carbon monoxide			
26.	Cracking	The reaction used in the oil industry to break down large hydrocarbon molecules into			
	0	smaller, more useful ones.			
27.	Thermal	The breaking down of a compound using heat.			
	decomposition				
28.	Rate of Reaction	Mean rate of reaction = $\frac{quantity of product formed}{time}$			
20	Equation	mantity of reactant yead			
29 .	Rate of Reaction	Mean rate of reaction = $\frac{4}{100000000000000000000000000000000000$			
30	Collision theory	An increased proportion of particles exceeding the activation energy has a greater effect			
50.		on rate than the increased frequency of collisions.			
		Week 4:			
31.	Bromine water test	Alkane: bromine water maintains its colour, so stays orange.			
		Alk <u>ene</u> : bromine water turns from orange to colourless.			
32.	Accuracy	Degree to which results of a measurement or calculation is close to the correct value.			

33.	Precision	Refinement in a measurement especially as represented by the number of digits given.	
34.	Resolution	The smallest interval measurable by a scientific instrument.	
35.	Significant figures	Representing a value to that number of digits, which are significant. For 3 Significant figures, examples include: 0.00435, 1.34, 4035, or 1.25	
36.	Decimal places	The number of digits you show after the decimal point, before rounding (if needed). For 3 decimal places, examples include: 0.234, 0.002 or 23.347.	
37.	Melting Point	The temperature at which it changes state from solid to liquid at atmospheric pressure.	
38.	Boiling Point	The temperature at which it changes state from liquid to gas at atmospheric pressure.	
39.	Pure substance	A substance which consists of just one type of atom or molecule.	
40.	Compound	A substance which contains two or more different elements in its structure.	
		Week 5:	
41.	Formulation	A mixture that has been designed for a useful product.	
42.	Nanoparticles	Very tiny particles or structures between 1-100 nanometres in size.	
43.	Rf (Retention factor)	A measurement from chromatography: it is the distance a spot of substance has been carried above the baseline divided by the distance of the solvent front. $Rf = \frac{distance \ moved \ by \ substance}{distance \ moved \ by \ solvent}$	
44.	Chromatography	The process whereby small amounts of dissolved substances are separated by running a solvent.	
45.	Sulphuric acid	H₂SO₄	
46.	Nitric acid	HNO ₃	
47.	Hydrochloric acid	нсі	
48.	Sulfate ion	SO ₄ ²⁻	
49.	Carbonate ion	CO ₃ ²⁻	
50.	Hydroxide ion	OH [.]	
		Week 6:	
51.	Ammonium ion	NH4 ⁺	
52.	Pipette	A glass tube used to measure accurate volumes of liquids.	
53.	Splint	A wooden stick which is ignited to carry a flame from one place to another.	
54.	Electrolysis	A process used to decompose a compound using electrical current.	
55.	Atmosphere	The relatively thin layer of gases that surround Earth.	
56.	Purification	A process used to move contaminants from a sample, to obtain a pure substance.	
57.	Waste water	Water which has come from the household and is dirty. For example the washing machine, tap and toilets.	
58.	Infrared spectroscopy	An analytical technique using infrared radiation to identify the bonds present in a substance.	
59.	Carbon footprint	The total amount of carbon dioxide and other greenhouse gases emitted over the full life cycle of a product, service or event.	
60.	Carbon capture	A technique where carbon dioxide produced from fossil fuel power stations is pumped	
	and storage	deep underground, to be absorbed by porous rocks.	
61	Bioleaching	A process used to extract copper using bacteria	
62.	Particulates	Small solid particles given off from motor vehicles as a result of incomplete combustion	
63.	Global dimming	Where particulates travel to the upper atmosphere and reflect sunlight back into space	
64.	Atmosphere	The relatively thin layer of gases that surround Earth.	
65.	Test for Oxygen	Relights a glowing splint.	
66.	Test for Carbon Dioxide	Turns limewater cloudy.	
67.	Test for Chlorine	Bleaches damp litmus paper.	
68.		Duran utility of forward up and	
•••	Test for Hydrogen	j Burns with a squeaky pop .	
69.	Test for Hydrogen Greenhouse Gases	Carbon dioxide, methane, water vapour	
69. 70.	Test for Hydrogen Greenhouse Gases Acid Rain	Carbon dioxide, methane, water vapour Caused by sulfur dioxide and nitrogen oxides.	

71.	Subatomic		relative charge	relative mass			
	Particles	proton	+1	1			
		neutron	0 neutral	1			
		electron	-1	1 / 1840			
72.	Transition element	Element from t	he central block c	of the periodic ta	ble.		
73.	Alkali Metals	Elements in gro	oup I of the Perio	dic Table– These	all have I electron in their outer shell.		
74.	Halogens	Elements in gro	oup 7 of the Perio	dic table– These	all have 7 electrons in their outer shell.		
75.	Ionic Bonding	The electrostat	tic force of attract	ion between pos	itively and negatively charged ions.		
76.	Giant ionic structures	A huge 3D latt	ice of lonically bor	nded atoms, e.g.	Sodium Chloride.		
77.	Covalent Bond	The bond betw	veen two or more	atoms that share	e one or more pairs of electrons.		
78.	Giant Covalent structures	A huge 3D net	work of covalently	y bonded atoms,	e.g. graphite		
79.	Oxidation	A reaction in w	hich oxygen is ad	ded or when ele	ctrons are lost.		
80.	Reduction	A reaction in w	A reaction in which oxygen is lost or electrons are gained.				
	Week 9:						
			Wee	e k 9:			
81	Rate of reaction	The rate at wh	Wee ich a reaction take be formed.	e k 9: es place, i.e for t	he reactants to be used up and the		
81 82	Rate of reaction Catalyst	The rate at wh products to b Substance whic alternative path	Wee ich a reaction take be formed. th speeds up a c l tway for a reaction	ek 9: es place, i.e for t hemical reaction n which has a low	he reactants to be used up and the on, without being used up. Provides an wer activation energy.		
81 82 83	Rate of reaction Catalyst Surface area	The rate at wh products to b Substance whic alternative path The total area than lumps of a	Wee ich a reaction take be formed. th speeds up a c l tway for a reaction of a substance acr a substance.	ek 9: es place, i.e for t hemical reaction n which has a low oss its whole sur	he reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area		
81 82 83 84	Rate of reaction Catalyst Surface area Concentration	The rate at wh products to b Substance whic alternative path The total area than lumps of a The amount of	Wee ich a reaction take be formed. th speeds up a c l way for a reaction of a substance acr a substance. of particles of a s	ek 9: es place, i.e for t hemical reaction n which has a low ross its whole sur substance in a ce	the reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area rtain volume.		
81 82 83 84 85	Rate of reaction Catalyst Surface area Concentration Alk <u>anes</u>	The rate at wh products to b Substance whic alternative path The total area than lumps of a The amount of A hydrocarbon group have a na	Wee ich a reaction take be formed. th speeds up a cl way for a reaction of a substance acr a substance. of particles of a s which contains o ame ending in -a n	ek 9: es place, i.e for t hemical reaction n which has a low ross its whole sur substance in a ce nly single carbo	the reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area rtain volume. on-carbon bonds. Molecules from this		
81 82 83 84 85 86	Rate of reaction Catalyst Surface area Concentration Alkanes	The rate at wh products to b Substance whic alternative path The total area than lumps of a The amount o A hydrocarbon group have a na A hydrocarbon structure. Mol	Wee ich a reaction take be formed. th speeds up a cl way for a reaction of a substance acr a substance. of particles of a s which contains o ame ending in -an which contains a ecules from this g	ek 9: es place, i.e for t hemical reaction n which has a low coss its whole sur substance in a ce nly single carbo ne. t least one dou roup have a nam	he reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area rtain volume. on-carbon bonds. Molecules from this ible carbon-carbon bond in its e ending in -ene.		
81 82 83 84 85 86 87	Rate of reactionCatalystSurface areaConcentrationAlkanesAlkenesSaturated	The rate at wh products to b Substance whic alternative path The total area than lumps of a The amount of A hydrocarbon group have a na A hydrocarbon structure. Mol A hydrocarbon	Wee ich a reaction take be formed. th speeds up a cl way for a reaction of a substance acr a substance. of particles of a s which contains o ame ending in -an which contains a ecules from this g which contains o	ek 9: es place, i.e for t hemical reaction n which has a low coss its whole sur substance in a ce nly single carbo t least one dou roup have a nam	the reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area rtain volume. on-carbon bonds. Molecules from this ible carbon-carbon bond in its e ending in -ene. on-carbon bonds.		
81 82 83 84 85 86 87 88	Rate of reactionCatalystSurface areaConcentrationAlkanesAlkenesSaturatedUnsaturated	The rate at wh products to b Substance whice alternative path The total area than lumps of a The amount of A hydrocarbon group have a na A hydrocarbon structure. Mol A hydrocarbon A hydrocarbon	Wee ich a reaction take be formed. th speeds up a cl way for a reaction of a substance acr a substance. of particles of a s which contains o ame ending in -an which contains a ecules from this g which contains o which contains o which contains o	ek 9: es place, i.e for t hemical reaction in which has a low coss its whole sur substance in a ce nly single carbo ine. t least one dou roup have a nam nly single carbo t least one dou	he reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area rtain volume. on-carbon bonds. Molecules from this ible carbon-carbon bond in its e ending in -ene. on-carbon bonds. ible carbon-carbon bond.		
81 82 83 84 85 86 86 87 88 88 89	Rate of reactionCatalystSurface areaConcentrationAlkanesAlkenesSaturatedUnsaturatedViscosity	The rate at wh products to b Substance whice alternative path The total area than lumps of a The amount of A hydrocarbon group have a na A hydrocarbon structure. Mol A hydrocarbon A hydrocarbon The resistance	Wee ich a reaction take be formed. th speeds up a cl way for a reaction of a substance acr a substance. of particles of a s which contains o ame ending in -an which contains a ecules from this g which contains o which contains o which contains a of a liquid to flow	ek 9: es place, i.e for t hemical reaction n which has a low coss its whole sur substance in a ce nly single carbo re. t least one dou roup have a nam nly single carbo t least one dou ring or pouring; a	the reactants to be used up and the on, without being used up. Provides an wer activation energy. face. A powder has a larger surface area rtain volume. on-carbon bonds. Molecules from this ible carbon-carbon bond in its e ending in -ene. on-carbon bonds. ible carbon-carbon bond. ible carbon-carbon bond.		

History Year II Unit I Germany, Conflict and Tension in Asia, Elizabeth & Health and the People

	Week I:		RAG
Ι.	What year did Germany become a unified country?	1871	
2.	What is the strong belief called Militarism?	A belief in strong armed forces	
3.	Who was the creator of the idea of the four humours?	Hippocrates	
4.	What were the four humours?	Blood, yellow bile, black bile and phlegm	
5.	What was often thought to be the cause of illness during the middle ages?	An imbalance of the four humours	
6.	Who was Elizabeth I's mother?	Anne Boleyn	
7.	What happened to her?	She was executed in 1536 (just before Elizabeth's third birthday).	
8.	What was the name of the US initiative to give Europe \$13 billion in 1947?	The Marshall Plan	
9.	Who said that an 'Iron Curtain' had divided Europe after 1945?	Winston Churchill	
10.	What was the Western military alliance called?	ΝΑΤΟ	
	Week 2:		
11.	The Kaiser ruled over German states. Each state had a what to represent them?	Bundesrat	
12.	The Kaiser was advised by a (state the title of the job)	Chancellor	
13.	What was the communist military alliance called?	The Warsaw Pact	
14.	What was Truman's policy of containing communism called?	The Truman Doctrine	
15.	Who was the Greek physician, surgeon and philosopher in the Roman Empire?	a) Galen	
16.	What were two tools that Medieval doctors used for diagnosing illness?	b) Urine charts c) Zodiac chars	
17.	What was a journey undertaken to a religious shrine, sometimes in an attempt to get forgiveness from God for your sins so that you might be healed, called?	Pilgrimage	
18.	Why was Elizabeth imprisoned in the Tower of London in 1554.	She was accused of supporting a rebellion against her sister Queen Mary.	
19.	What year did Elizabeth I become Queen of England?	1558	
20.	How old was she when she was coronated?	25	
	Week 3:		
21.	What were two common treatments used to balance the four humours?	a) Purging b) Bleeding	
22.	Name two Muslim doctors who had a great influence on Western medicine.	Rhazes and Avicenna	
23.	During the Middle Ages what did the Church recommend as the best treatment for illness?	Prayer	
24.	What was the name for Kaiser Wilhelm's world policy?	Weltpolitik	
25.	What was made up of the House of Lords and the House of Commons and influenced tax and was responsible for passing laws?	Parliament	

26.	Why did it have limited power during Elizabeth's reign?	She decided when to call Parliament and how much of their advice she should listen to.	
27.	Who were Elizabeth's main advisors who were responsible for the day to day running of the country?	The Privy Council	
28.	Who were appointed by the queen to take administrative responsibility for a particular area of the country.	Lord Lieutenants	
29.	Kaiser Wilhelm II introduced to Germany	industrialisation	
30.	What is the line which separates North Korea from South Korea called?	38 th parallel	
	Week 4:		
31.	What was the name of the leader of North Korea in 1948?	Kim-II-sung	
32.	What was the name of the leader of South Korea in 1948?	Syngman Rhee	
33.	What is the capital of South Korea?	Seoul	1
34.	What was the name of the new political party which many ordinary workers voted for during the reign of Kaiser Wilhelm II?	Social Democratic Party (SPD)	
35.	Who's main role was to ensure that the laws passed by Parliament were properly enforced?	Justices of the Peace (JPs)	
36.	Who was Elizabeth I's most trusted advisor and served as Secretary of State twice?	William Cecil	
37.	Who was Elizabth's 'spymaster'?	Francis Walsingham	
38.	What happened in 1562?	Elizabeth nearly died from smallpox	
39.	What religion was Elizabeth I?	Protestant	
40.	Who was Elizabeth's childhood friend, who many assumed she would marry.	Robert Dudley	
	Week 5:		
41.	What did a series of Naval Laws introduced between 1898 and 1912 allow Germany to achieve?	Germany to build a navy as big as Britain's navy.	
42.	What 3 key impacts did World War I have on Germany?	a) Bankrupt b) Politically unstable c) Defeated	
43.	What does the term capitalist mean?	Individuals are free to own businesses and land and compete with others.	
44.	What is the technique called when you apply heat to a wound to stop the blood flowing?	cauterisation	
45.	When was the Northern Rebellion, led by Catholic lords in England?	1569	
46.	What rebellion did Elizabeth face in 1571?	The Ridolfi Plot	
47.	Who was executed in 1572 for his involvement in the Northern Rebellion and the Ridolfi Plot?	The Duke of Norfolk	
48.	Who did Elizabeth make the Lord Lieutenant of Ireland in I 599?	Robert Devereux (The Earl of Essex)	
49.	In what year did he lead a rebellion against Elizabeth for which he was executed?	1601	

50.	On what idea was Elizabethan society based, with God at the top followed by angels, human beings, animals and plants?	The Great Chain of Being	
	Week 6:		
51.	What position (job title) was Friedrich Ebert, the leader of the SPD, the first person to have?	Germany's first President	
52.	Ebert changed Germany from a monarchy to having no monarchy. What was this called?	Weimar Republic	
53.	What was the name of the US Commander-in-Chief of UN forces in Korea?	General MacArthur	
54.	What was the reason for MacArthur's sacking?	For sending troops back into North Korea	
55.	What were the three groups underneath the Queen in Elizabethan society?	the nobility, followed by the gentry and then the Peasantry.	
56.	How did rich Elizabethans show off their wealth through their food?	Having meals made up mostly of meat and drinking wine.	
57.	What fashion trend did wealthy Elizabethan women follow?	They whitened their faces and wore fine clothes.	
58.	What was the name of the book written by the famous Medieval surgeon John Ardene in 1350?	The Practice of Surgery	
59.	What is a place to put human waste called?	cesspit	
60.	Name an Arab Muslim physician and surgeon who invented 26 new surgical instruments.	Abulcasis	
	Week 7:		
61.	What was the Weimar constitution?	A formal set of rules of how Germany was governed.	
62.	Give one example of a rule from the constitution.	Everyone over 20 could vote	
63.	What was the name used for Vietnam, Laos and Cambodia before the Second World War and up to the Battle of Dien Bien Phu?	French Indochina	
64.	Why did they do this?	To show they did not have to work outside and get a tanned face.	
65.	What was a key element of both men and women's fashion?	The ruff worn around the neck.	
66.	What did rich Elizabethans' build to show off their wealth?	Great country houses	
67.	What was one of the most popular forms of Elizabethan entertainment?	Theatre	
68.	What did people think was a major cause of disease in the Middle Ages and Renaissance period (not God or the four humours this time)?	Miasma (bad smells in the air)	
69.	When did the Black Death arrive in England?	1348	
70.	What caused this disease to spread?	Infected fleas carried by rats.	
	Week 8		
71.	What was the voting system used for the Reichstag (German Parliament) called?	Proportional Representation	
72.	What was the position (job title) directly beneath the President within the German Government?	Chancellor	
73.	What was the country called which controlled Indochina during the Second World War?	Japan	

74.	Who led the Vietnamese communists fighting the French after the Second World War?	Ho Chi Minh	
75.	What was he famous for as a military leader.	Using Guerilla tactics to defeat the French.	
76.	Which playwright was particularly popular during Elizabeth's reign?	William Shakespeare	
77.	Which theatre was built by the Lord Chamberlain's men (William Shakespeare's theatre company) in 1599?	The Globe Theatre	
78.	Which religious group was opposed to the theatre?	Puritans	
79.	List two consequences for the Black Death.	a) Food shortages b) Rising prices	
80.	What did the peasants demand following the Black Death and why?	Higher wages as there were less workers	
	Week 9:		
81.	What were the Germans forced to sign in 1919?	The Treaty of Versailles	
82.	What was the army reduced to as a result of the Treaty of Versailles?	100, 000	
83.	In what battle were the French were decisively beaten by the Vietnamese communists?	The Battle of Dien Bien Phu	
84.	What was the name of the conference that was to decide the future of Vietnam after ths battle?	The Geneva Conference	
85.	What was the myth in which Hitler and other nationalists blamed groups such as the Communists and Jews for Germany's defeat in WWI?	The stab in the back myth	
86.	What is Elizabeth's reign often referred to?	a 'golden age'	
87.	Why was this?	It was an era of new scientific experimentation, technological development and new ideas in the arts.	
88.	What was alchemy?	Trying to turn cheap metal into gold	
89.	When was the Middle Ages (give dates)	1000-1500	
90.	What is the cutting open of a body or plant to study it called?	Dissection	

French - Year II - Revision of tenses, justifications and negatives

	Week I: Present tense Le présent		RAG
1.	I have	J'ai	
2.	l am	Je suis	
3.	l do/make	Je fais	
4.	l go/am going	Je vais	
5.	l play/am playing	Je joue	
6.	I have to (help)	Je dois (aider)	
7.	l can (sing)	Je peux (chanter)	
8.	l want (to work)	Je veux (travailler)	
9.	l prefer	le préfère	
10.	I think that	le pense que	
	Week 2: The perfect tense Le passé co	mposé	
11.	I have been	J'ai été	
12.	l went	Je suis allé(e)	
13.	I have done / I did	J'ai fait	
14.	I have watched	J'ai regardé	
15.	I have drunk / I drank	J'ai bu	
16.	I have read / I read	J'ai lu	
17.	I have had / I had	J'ai eu	
18.	I have worked / I worked	J'ai travaillé	
19.	I have preferred / I preferred	J'ai préféré	
20.	I have thought that / I thought that	J'ai pense que	
21	Lused to play / Lwas playing		
21.	Lused to be / Lwas	Pótais	
23.	Lused to have / I had	l'avais	
24.	I used to watch / I was watching	Je regardais	
25.	l used to go	l'allais	
26.	l used to eat	Je mangeais	
27.	l used to sing	Je chantais	
28.	l used to work	Je travaillais	
29.	l used to prefer	Je préférais	
30.	l used to think that	Je pensais que	
	Week 4: The near future Le futur procl		
31.	I am going to have	Je vais avoir	
32.	I am going to be	Je vais être	
33.	I am going to go	Je vais aller	
34.	I am going to do	Je vais faire	
35.	I am going to watch	Je vais regarder	
36.	I am going to listen	Je vais écouter	
37.	I am going to sing	Je vais chanter	
38.	I am going to work	Je vais travailler	
39.	I am going to prefer	Je vais préférer	
40.	I am going to think	Je vais penser	
	Week 5: The future Le futur		
41.	I will have	J'aurai	
42.	I will be	Je serai	
45.	I will do	j irai le ferzi	
45.	l will watch	le regarderaj	
46	l will listen	l'écouterai	
L'.			

47.	I will sing	Je chanterai	
48.	l will work	Je travaillerai	
49.	l will prefer	Je préférerai	
50.	l will think	le penserai	
	Week 6: The conditional Le condition		
51.	I would have	l'aurais	
52.	I would be	le serais	
53.	I would go	l'irais	
54.	I would do	Je ferais	
55.	I would watch	Je regarderais	
56.	I would like	Je voudrais / J'aimerais	
57.	I would sing	Je chanterais	
58.	l would work	Je travaillerais	
59.	l would prefer	Je préférerais	
60.	I would think	le penserais	
	Week 7: Le plus-que-parfait The pluperfec	t	
61.	Before going on holiday, I had bought new clothes	Avant de partir en vacances, j'avais acheté des nouveaux vêtements	
62.	Unfortunately, I had eaten all the sweets	Malheureusement, j'avais mangé tous les bonbons	
63.	Before going out, my friend had done her homework	Avant de sortir, ma copine avait fait ses devoirs	
64.	I had tried!	J'avais essayé!	
65.	I had recycled the paper	J'avais recyclé le papier	
66.	We had reduced the pollution	On avait réduit la pollution	
67.	I had gone to the bakery	J'étais allé(e) à la boulangerie	
68.	She had gone to university	Elle était allé(e) à l'université	
69.	He had gone on holiday	Il était parti en vacances	
70.	They had returned to England	Ils étaient retournés en Angleterre	
	Week 8: The subjunctive Le subjonctif		
71.	The following expressions are followed by the subjunctive:		
72.	before	avant que	
73.	although	bien que	
74.	provided that	à condition que	
75.	we / you / one must / it is necessary that	il faut que	
76.	I have a good job	j'aie un bon travail	
77.	I make an effort	je fasse un effort	
78.	my results to be good	mes résultats soient bons	
79.	I am disciplined	je sois disciplinė(e)	
80.	I go to university	j'aille à l'université	
01	week 9: Justifications and negatives		
01.	because	parce que / car	
82	because	va que / puisque	
84	l don't want		
85		le ne fume jamais	
86	He doesn't do anything	Il ne fait rien	
_ 00.			
87	l don't see anybody	le ne vois personne	
87. 88.	I don't see anybody There is no more homework	Je ne vois personne Il n' y a plus de devoirs	
87. 88. 89.	I don't see anybody There is no more homework	Je ne vois personne Il n'y a plus de devoirs le n' ai gu' une soeur	

	Week I: RAG		
Ι.	Natural Hazard	A natural event that threatens people or has the potential to cause damage, destruction and death.	
2.	Destructive Plate Margin example	The Pacific Ring of Fire.	
3.	Constructive plate margin example	The Mid Atlantic Ridge.	
4.	Conservative plate margin example	The San Andreas Fault.	
5.	L'Aquila 2009 size, date and time.	Size: Magnitude 6.3 on the Richter scale Year: 2009 Time: 3:32 am	
6.	L'Aquila 2009 primary effects	308 people died / 1,500 injured / 10-15000 buildings collapsed, San Salvatore hospital damaged, many historical buildings damaged.	
7.	L'Aquila 2009 secondary effects	Landslides triggered by aftershocks damaged roads / number of students at L'Aquila university decreased / rents on housing increased.	
8.	L'Aquila 2009 immediate responses	40, 000 tents used for shelter / the Italian red cross searched for survivors within one hour / water and food distributed / mortgages and bills were suspended / EU gave \$552.9 million to begin rebuilding.	
9.	L'Aquila 2009 Long term responses	Residents did not pay taxes in 2010 during the immediate recovery period / students were given free tuition fees to attend the university / I government official and 6 scientists went to prison during investigations / it took approximately 15 years to rebuild L'Aquila.	
10.	Nepal 2015 size, date and time.	Size: 7.8 on the Richter scale Year: 2015 Time: 11:56 am	
		Week 2:	
11.	Nepal 2015 primary effects	8,841 people died / 16,800 injured / 26 hospitals, and 50% of schools destroyed, reduced supply of food, water and electricity.	
12.	Nepal 2015 secondary effects	I million people homeless / avalanche on Mount Everest killed 19 people / reduced numbers of tourists visiting Nepal / rice seed was destroyed meaning people could not grow food.	
13.	Nepal 2015 immediate responses	Nepal requested international help / the UK gave \$126 million for emergency aid / the Red Cross provided tents for 225,000 people / the WHO distributed medical supplies / sherpas carried supplies into hard to reach areas / Facebook launched its Safe feature.	
14.	Nepal 2015 long term responses	Nepal created a Post-Disaster Needs Assessment / 23 areas needed completely rebuilding / \$274 million was promised from abroad for rebuilding / Mount Everest was opened by August for tourists.	
15.	Why do people live in Hazard zones?	Geothermal energy (Iceland) / Farming (ash creates fertile soil) / Tourism (volcanoes are beautiful) / Poverty (attachments to the area you are from).	
16.	Typhoon Haiyan 2013 size and date.	Size: Category 5 with wind speeds up to 314km/hr Date: hit the Philippines on 8 November 2013.	
17.	Typhoon Haiyan 2013 primary effects	6190 people died / 90% of Tacloban city destroyed / airport, homes and roads badly damaged / seed stocks lost.	
18.	Typhoon Haiyan 2013 secondary effects	Oil barges ran aground causing a massive oil leak / looting happened as survivors fought for food supplies / sea water contaminated the land	
19.	Typhoon Haiyan 2013	800,000 people were evacuated before the storm / aid arrived 3 days later by plane /	

	immediate responses	curfew was imposed to stop looting / \$1.5 billion of foreign aid was pledged / main airport reopened after 3 days / power was restored after a week.	
20.	Typhoon Haiyan 2013 long term responses	Build Back Better was the pledge of the government (no build areas designated along some coastlines) / new storm surge warning system / mangroves replanted.	
		Week 3:	
21.	Reducing the effects of tropical storms.	Monitoring - satellites and planes to see storms emerging, improvements in prediction using computer models.	
		Protection - storm shelters, shutters on windows, better housing, remove trees that are close to buildings.	
		Planning - disaster supply kits, training emergency services, building evacuation centres.	
22.	Somerset levels 2014 causes	Flooding of the rivers Tone and Parrett - this was because it was the wettest January since records began. High tides prevented water getting to the sea. Lastly, the rivers had not been dredged for 20 years.	
23.	Somerset levels 2014 effects	Social - 600 homes flooded, 16 farms evacuated, villages cut off, power supplies down. Economic - £10 million in damage, farmers lost money, people could not get to work. Environmental - sewage contaminated flood water debris left from the flood.	
24.	Somerset levels 2014 responses	Immediate responses - Homeowners coped the best they could, Villages cut off used boats to get shopping etc, Local volunteers helped out. Long-term - $\pounds 20$ million flood action plan by the environment agency, 8km of the Tone and Parrett were dredged, river banks and roads have been raised, some flood defences built.	
25.	Evidence of climate change	Ice cores (CO2), tree rings, historical records such as diaries, current sea level rise and visible melting of the ice caps.	
26.	Causes of natural climate change	I) volcanic eruptions (global cooling) 2) orbital changes/Milankovitch cycles 3) solar activity.	
27.	The greenhouse effect	The trapping of the sun's heat by gas in the atmosphere.	
28.	The human causes of climate change	Increase in carbon dioxide due to burning of fossil fuels in power stations and cars. Increase in methane production due to livestock and rice farming.	
29.	Adaptation to climate change examples	Maldives - raising houses on stilts, sea defences, restoring mangroves. Himalayas - creating artificial glaciers to store water until the summer. The Gambia - shade trees planted, new efficient irrigation systems, drought resistant crops used.	
30.	Mitigation of climate change examples	Carbon capture, renewable/green energy sources, planting trees, international agreements to cut greenhouse gas emissions.	
		Week 4	
31.	Destructive and constructive waves	Destructive waves have a bigger swash than backwash, a larger wave height and a shorter wavelength, they erode the beach. Opposite for constructive waves.	
32.	Headlands and bays	Formed when there are bands of hard and soft rock lying at an angle to the sea. The softer rock erodes faster creating bays. The harder rack is left sticking out to sea forming headlands.	
33.	Spits	Spits are formed where there is a large amount of eroded material that is moved along the shore by longshore drift. If the coastline changes direction at a river mouth then the sea may lose energy, depositing the material forming a spit. Spits have a hooked end as they cannot grow across a river and salt marshes grow up behind spits due to mud trapped from the river.	
34.	Sea walls	Reflect wave energy, they are very effective but very expensive.	
35.	Groynes	Trap sediment building up the beach, this is great for tourism but speeds up erosion along the coast.	

36.	Rock armour	Absorbs wave energy, these are very effective and cheaper than a sea wall but they are still expensive and make getting to the beach difficult.	
37.	Gabions	Absorbs wave energy and allow cliff drainage, these are cheap but break easily.	
38.	Dune regeneration	Planting marram grass to trap sand building up the dunes, great for wildlife but dunes will not stand up to big storms.	
39.	Dune fencing	Fences trap sand building up the dunes, they also keep people off of the dunes.	
40.	Beach nourishment	Building up the beach to act as a barrier, this will need repeating often.	
		Week 5	
41.	Lyme Regis		
42.	Lyme Regis		
43.	Lyme Regis		
44.	Gorges	Gorges form in the upper course of a river when hard rock lies over soft rock. At a waterfall the water falls into the plunge pool and the energy spreads out causing the soft rock to erode creating an overhang, Eventually the overhang collapses and the waterfall retreats leaving a gorge.	
45.	Oxbow lakes	Oxbow lakes occur in the middle course where meanders move; this is because the water moves faster on the outside of a meander causing erosion, deposition occurs on the inside where the water is slower. Over time meanders may move towards each other. When they reach each other the river takes the shortest route, reducing the seed of water on the meander, this causes deposition to occur cutting off the old meander, forming an oxbow lake.	
46.	Floodplains and Levees	Floodplains occur in the lower course when menders move sideways eroding valley sides. Levees are formed when rivers flood, sediment is deposited on the side of the river forming a river bank.	
47.	Human causes of Flooding	Urbanisation, deforestation, soil compaction by machinery.	
48.	Physical causes of flooding.	Amount of precipitation, saturated soil, impermeable rock, snow melt, steep slopes.	
49.	Flood management	Hard engineering - Dams, Embankments, Flood relief channels, channel straightening. Soft engineering - Planting trees, floodplain zoning, river restoration, warning systems.	
50.	Boscastle flood management	Aim - to prevent Boscastle from the risk of another flash flood occurring Strategies - raised arch or bridged, channel widened, land-use zoning, gauging station to monitor and predict, flood wall, clearing vegetation to avoid debris/blockages.	
		Week 6	
51.	Natural Hazard	A natural event that threatens people or has the potential to cause damage, destruction and death.	
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